## Synthesis of an Enol-ether of a Cyclopropanone from a Diazoalkenylether: a Novel Class of Compound

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Summary Irradiation of 1-diazo-2-methoxy-3-methyl-1phenylbut-2-ene (4), prepared from the p-toluenesulphonylhydrazones (3) of 2-methoxy-3-methyl-1-phenylbut-2-en-1-one (2), yields 1-methoxy-3,3-dimethyl-2phenylcyclopropene (5).

ENOL derivatives of cyclopropanones are potential intermediates of enolate anions of cyclopropanones1 and of bicyclobutanes<sup>2</sup> with angular oxygen functions.<sup>3</sup> The synthesis of 1-methoxy-3,3-dimethyl-2-phenylcyclopropene (5), the first example of such compounds, has been reported<sup>4</sup> but it was prepared in disappointingly low yield. We now describe a more convenient synthesis of this compound.

Closs et al.<sup>5</sup> have reported a convenient synthesis of alkylcyclopropenes from diazoalkenes. We now report the synthesis of a diazoalkenylether which on irradiation undergoes smooth transformation to 1-methoxy-3,3-dimethyl-2-phenylcyclopropene (5). This route is also potentially a more general pathway† to these compounds than that described earlier.4

Stirring the lithium salt<sup>6</sup> of the acid<sup>7</sup> (1) with phenyllithium in ether at room temperature for 48 h gave 2methoxy-3-methyl-1-phenylbut-2-en-1-one (2) b.p. 76°, 1.0 mmHg (73%). On treatment with *p*-toluenesulphonylhydrazine in methanol at  $40^{\circ}$  for 14 days the ketone (2) gave a mixture of the geometrically isomeric p-toluene sulphonylhydrazones (3) in virtually quantitative yield. Stirring an acid solution of (3) for several days at room temperature caused conversion to only one isomer, m.p. 145—146° (decomp.). Pyrolysis<sup>8</sup> of the dry lithium salts of (3) (170°, 0.7 mmHg) gave 1-diazo-2-methoxy-3-methyl-1phenylbut-2-ene (4) as a liquid (70%) [i.r. (film) v 2050, 1650 cm<sup>-1</sup>; n.m.r. (CCl<sub>4</sub>)  $\delta$  6.87–7.33 (5H, m), 3.40 (3H, s), 1.83 (3H, s), 1.63 (3H, s)]. Irradiation (Philips 125 W) of a 3% solution (CCl<sub>4</sub>) of compound (4) in a pyrex vessel at room temperature for 40 min gave a solution containing

only 1-methoxy-3,3-dimethyl-2-phenylcyclopropane (5) as shown by its n.m.r. spectrum<sup>4</sup> (CCl<sub>4</sub>),  $\delta$  6.93-7.33 (5H, m), 3.97 (3H, s), 1.37 (6H, s). Complete evaporation of the solvent left an oil, the i.r. spectrum of which was identical



to that obtained previously.<sup>4</sup> Of note is the strong absorpion<sup>9</sup> at 1840 cm<sup>-1</sup>. The neat liquid appeared to polymerise readily but solutions could be stored at low temperature for extended periods.

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† Preliminary work shows that compounds (6) and (7) can be synthesised by an analogous sequence.

- <sup>1</sup> J. Ciabattoni, P. J. Kocienski, and G. Melloni, Tetrahedron Letters, 1969, 1883.
- <sup>2</sup> P. G. Gassman and W. J. Greenlee, *J. Amer. Chem. Soc.*, 1973, 95, 980. <sup>3</sup> R. W. Sinclair, Ph.D. Thesis, Adelaide, 1970.

- <sup>4</sup> D. P. G. Hamon and L. J. Holding, *Chem. Comm.*, 1970, 1330. <sup>5</sup> G. L. Closs, L. E. Closs, and W. A. Böll, *J. Amer. Chem. Soc.*, 1963, 85, 3796.
- <sup>6</sup> T. M. Bare and H. O. House, Org. Syntheses, 1969, **49**, 81. <sup>7</sup> L. N. Owen, J. Chem. Soc., 1949, 3089.
- <sup>8</sup>G. M. Kaufman, J. A. Smith, G. G. Vander Stouw, and H. Shechter, J. Amer. Chem. Soc., 1965, 87, 935.
- \* 1840 cm<sup>-1</sup> is the expected region for absorption of a fully substituted cyclopropene: G. L. Closs, Adv. Alicyclic Chem., 1966, 1, 70.